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W9.1 (a) 1 mole of Li atoms = 7 g. (b) 1 mole

of Na atoms = 23 g. (c) 1 mole of Ca atoms =

40 g. (d) 1 mole of Fe atoms = 56 g. (e) 1

mole of Ag atoms = 108 g. (f) 1 mole of Pb

atoms = 207 g. W9.2 (a) 1 mole of Cl atoms =

35.5 g 10 moles of Cl atoms = 35.5 × 10 =

355 g.

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Answers Author: thebrewstercarriageho

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OpenStax. 1. The cutting edge of a knife
that has been sharpened has a smaller
surface area than a dull knife. Since
pressure is force per unit area, a sharp
knife will exert a higher pressure with the
same amount of force and cut through
material more effectively. 3.

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and types of bonding. The most interesting
question is why noble gases cannot form a
chemical bond? Because that is stable.
Every person wants to be rich, the same
goes for the atom too. [lock]

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