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Chapter 9 Practice Test -

March, 26 2023



Naming and Writing Chemical Formulas

Chapter 9 Practice Test - Naming and Writing Chemical Formulas Answer Section

MATCHING 1. ANS: A PTS: 1 DIF: L1 REF: p. 253 OBJ:

9.1.1 Identify the charges of monatomic ions by using the periodic table, and name the ions. STA: Ch.5.a 2.ANS: H

PTS: 1 DIF: L1 REF: p. 254

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This chapter is very important chapters for your chemistry notes for class 9. This chapter is about the bonding of the atoms and types of bonding. The most interesting question is why noble gases cannot form a chemical bond? Because that is stable. Every

person wants to be rich, the same goes for the atom too. [lock]

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formula. 2. Use preprefixes appropriately to indicate the number of each kind of atom. 3. End the name of the second element with suffix -ide

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Chapter 9 - Ionic and Covalent
Bonding . 1. In which pair do both
compounds exhibit predominantly
ionic bonding? A) RbCl and CaO
B) PCI5 and HF C) KI and O3 D)
Na2SO3 and BH3 E) NaF and
H2O . ANS: A PTS: 1 DIF: easy
REF: 9.1 OBJ: Define ionic bond.

TOP: bonding | ionic bonding
MSC: general chemistry . 2.
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substances in the reaction B. The energy released in the reaction C. The electron configuration of all all ...

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